

Oligonucleotide Analogues with a Nucleobase-Including Backbone:

Part 6

2-Deoxy-D-erythrose-Derived Phosphoramidites: Synthesis and Incorporation into 14-Mer DNA Strands

by Werngard Czechtizky and Andrea Vasella*

Laboratorium für Organische Chemie, ETH-Zentrum, Universitätstrasse 16, CH-8092 Zürich

Two modified DNA 14-mers have been prepared, containing either a 2-deoxy-D-erythrose-derived adenosine analogue carrying a C(8)–CH₂O group (dA*), or a 2-deoxy-D-erythrose-derived uridine analogue, possessing a C(6)–CH₂O group (dU*). These nucleosides are linked *via* a phosphinato group between O–C(3') (dA* and dU*) and O–C(5') of one neighbouring nucleotide, and between C(8)–CH₂O (dA*), or C(6)–CH₂O (dU*) and O–C(3') of the second neighbour. *N*⁶-Benzoyl-9-(β-D-erythrofuransyl)adenine (**3**) and 1-(β-D-erythrofuransyl)uracil (**4**) were prepared from D-glucose, deoxygenated at C(2'), and converted into the required phosphoramidites **1** and **2**. The modified tetradecamers **31** and **32** were prepared by solid-phase synthesis. Pairing studies show a decrease in the melting temperature of 7 to 8 degrees for the duplexes **31**·**30** and **32**·**29**, as compared to the unmodified DNA duplex **29**·**30**. A comparison with the pairing properties of tetradecamers similarly incorporating deoxyribose- instead of the deoxyerythrose-derived nucleotides evidences that the CH₂OH substituent at C(4') has no significant effect on the pairing.

Introduction. – Oligonucleotide analogues with a nucleobase-including backbone should allow to answer the question whether the structural differentiation between nucleobase and backbone in DNA, RNA, and their analogues is a necessary prerequisite for the formation of stable homo- and/or heteroduplexes [1–5]. We have already described oligonucleotide analogues with a phosphinato group between O–C(3') and either a C(8)–CH₂O group of 2'-deoxyadenosine (dA_n*¹); *Fig. 1*), or a C(6)–CH₂O group of 2'-deoxyuridine (dU_n*¹); *Fig. 1*) [5]. Force-field calculations [6] and *Maruzen* model studies suggested that these analogues may form autonomous pairing systems, and that the incorporation of single modified dA* or dU* units into DNA 14-mers is compatible with duplex formation. Pairing studies of singly modified tetradecamers, however, showed that the introduction of dA* or dU* units into a 14-mer DNA duplex lowered the melting temperature by 6–7°, as compared to the unmodified DNA duplex [5]. This destabilization may be due to a distortion of the DNA duplex caused by the *syn*-conformation [7] of the dA* or dU* units, as enforced by an unfavourable interaction between the C(8)- or C(6)-substituent and the carbohydrate moiety – presumably the O–C(5') group – of dA* and dU*. To reduce this steric interaction, we decided to remove the CH₂OH substituent at C(4') of dA*

¹) The repeating units of B-DNA are denoted dA, dT, dG, and dC. 2-Deoxyribose-derived units linked *via* O–C(3') and O(10), or O(7) are denoted dA* and dU*, respectively, and 2-deoxy-D-erythrose-derived units linked *via* O–C(3') and O(10), or O(7), dA* and dU*, respectively (*Fig. 1*).

and dU*, and to introduce the 2-deoxyerythrose-derived units deA*⁽¹⁾ and deU*⁽¹⁾ (Fig. 1) into a DNA duplex.

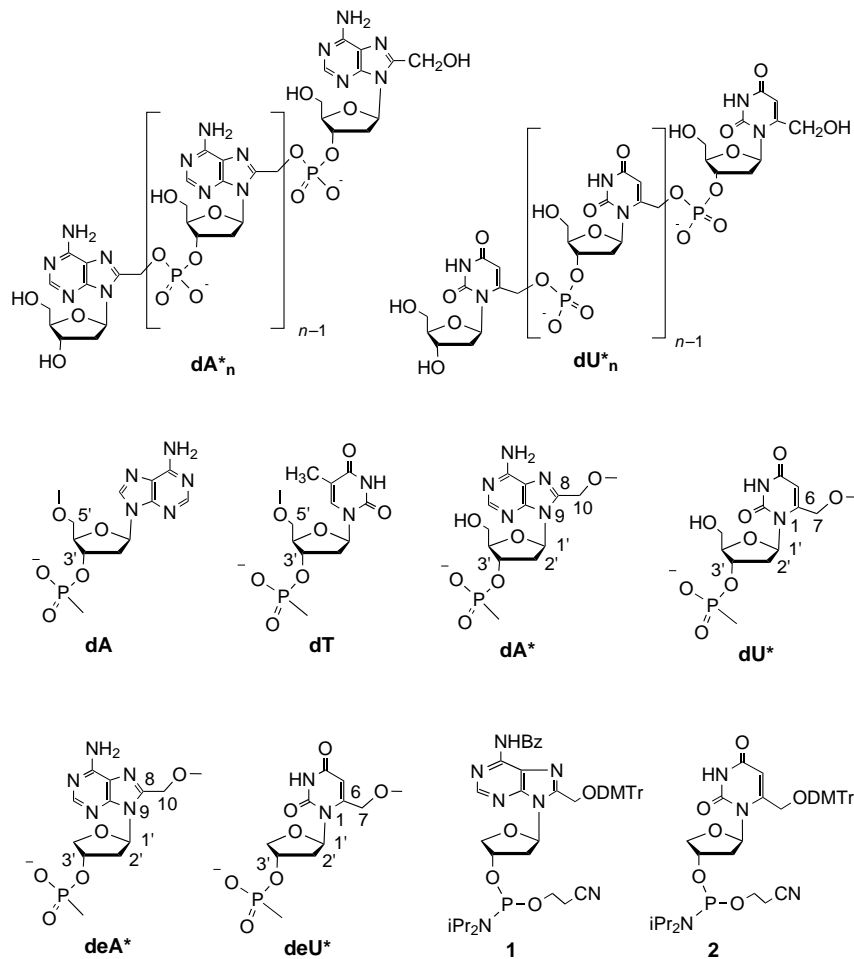


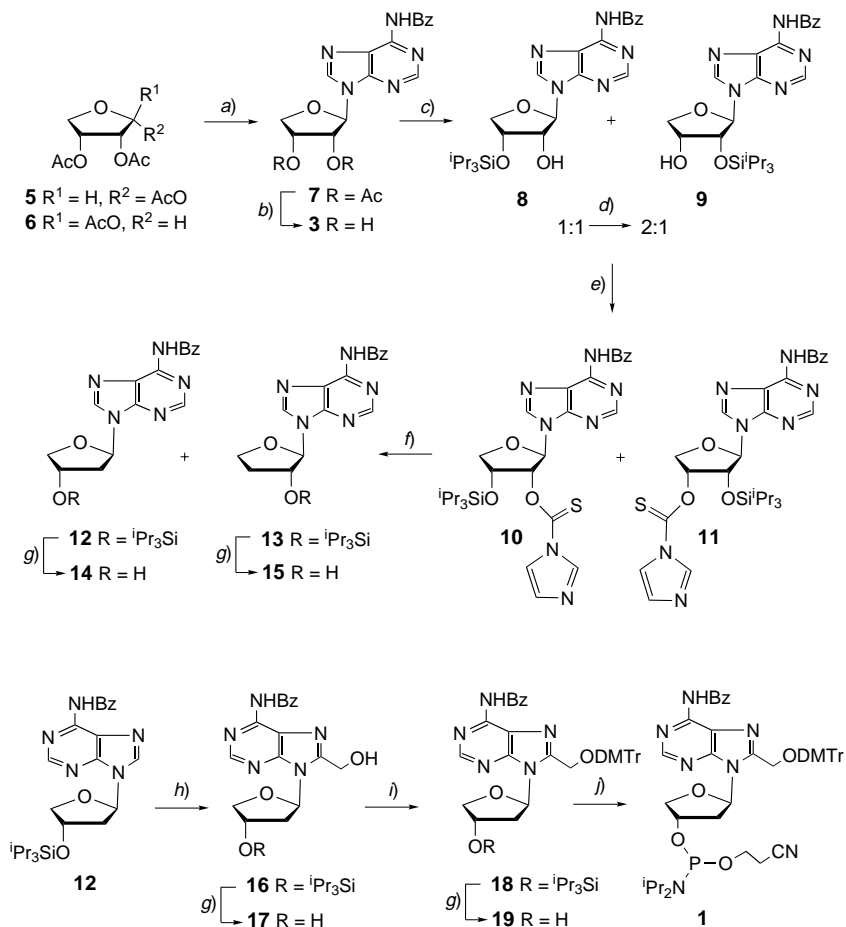
Fig. 1. 2-Deoxyribose-derived oligonucleotide analogues dA_n^* and dU_n^* , DNA units dA and dT, 2-deoxyribose-derived units dA* and dU*, 2-deoxyerythrose-derived units deA* and deU*, and the phosphoramidites **1** and **2**

We report on the syntheses of the deA*- and deU*-derived phosphoramidites **1** and **2** (Fig. 1), the incorporation of a single deA*, or deU* unit into a tetradecamer, and on the pairing of these tetradecamers with an unmodified complementary strand.

Results and Discussion. – 1. *Synthesis of the Phosphoramidites 1 and 2.* Nucleosidation of 2-deoxy-D-erythrose derivatives with thymine [8] and uracil [9] proceeds with a low selectivity, affording 2:1 to 1:1 mixtures of anomers, while nucleosidation of anomeric mixtures of triacetyl erythrofuranose with *N*⁶-benzoyladenine or uracil afforded diastereoselectively the β -D-nucleosides **3** and **4** (Schemes 1 and 2) after deprotection [10]. For this reason, we decided to prepare the phosphoramidites

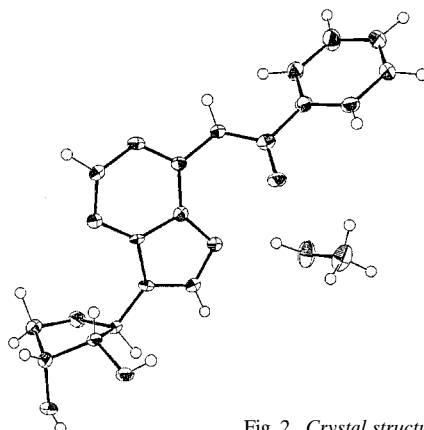
1 and **2** from the 1,2,3-tri-*O*-acetyl- α,β -D-erythroses **5/6** (Scheme 1) rather than from 2-deoxy-D-erythrose, and to subsequently remove the OH group at C(2'). Triacetyl-erythrose is readily accessible by degradation of D-glucose with Pb(OAc)₄ [11][12] and acetylation [13]²). Deoxygenation of **3** and **4**, followed by hydroxymethylation, protection, and phosphorylation, should lead to the phosphoramidites **1** and **2**.

Scheme 1



a) 1. *N*⁶-Benzoyladenine, *N,O*-bis(trimethylsilyl)acetamide (BSA); 2. SnCl₄; 80%. b) 1. NaOH, H₂O/MeOH/THF 5:4:1; 2. NH₄Cl; 80%. c) ⁱPr₃SiOTf, pyridine, DMF; 92%. d) SiO₂, Et₃N. e) *N,N*-(thiocarbonyl)diimidazole, CH₂Cl₂. f) Bu₃SnH, 2,2'-azobis(isobutyronitrile) (AIBN), toluene; 44% of **12** and 20% of **13** (from **8/9** 2:1). g) Bu₄NF · 3 H₂O, THF; **14** (74%); **15** (89%); **19** (76%). h) 1. LDA/THF; 2. DMF; 3. AcOH; 4. NaBH₄/EtOH; 88%. i) Dimethoxytrityl chloride (= bis(4-methoxyphenyl)(phenyl)methyl chloride; DMTrCl), EtNⁱPr₂, 4-(dimethylamino)pyridine (DMAP), CH₂Cl₂; 89%. j) (NCCH₂CH₂O)P(ⁱPr₂N)Cl, CH₂Cl₂, EtNⁱPr₂; 80% of 2 diastereoisomers (1:1).

2) 2-Deoxy-D-erythrose has been prepared from 3-deoxy-D-erythropentose [14], dimethyl (*S*)-malate [15], D-isoscorbic acid [16], and (*S*)-solketal [16].

Fig. 2. Crystal structure of **3**·MeOH

Nucleosidation of the anomeric mixture **5/6** with *N*⁶-benzoyladenine in the presence of SnCl₄ and *N,O*-bis(trimethylsilyl)acetamide (BSA) at 60° (*cf.* [17][18]) gave the β-D-configured nucleoside **7** (80%), which was deacetylated with 2M NaOH to yield 80% of the nucleoside **3**. Stannylation of **3** with Bu₂SnO in toluene [19], followed by treatment of the crude Sn-acetal with Et₃SiCl, led to a 1:1 mixture of regioisomeric silyl ethers. Regioselective silylation of **3** at HO–C(3') with various trialkylchlorosilanes in the presence of Ag salts and DABCO as reported for ribonucleosides [20] led again to unsatisfactory results. Finally, we treated **3** with a 2.5-fold excess of ⁱPr₃SiOTf in DMF/pyridine 16:1 at 23° to obtain 92% of a 1:1 mixture of the 3'- and 2'-O-ⁱPr₃Si-protected nucleosides **8** and **9**. Treatment of this mixture with silica gel in Et₃N led to partial migration of the ⁱPr₃Si group from O(2') to O(3'), resulting in a 2:1 mixture **8/9** after 24 h. This mixture was converted to the thiocarbamates **10/11** (2:1). Treatment of the crude mixture with Bu₃SnH and 2,2'-azobis(isobutyronitrile) (AIBN) in refluxing toluene yielded a mixture **12/13** (2:1; 64% from **8/9**). The nucleosides **12/13** were readily separated by FC. Desilylation of **12** and **13** yielded **14** and **15** (80% each), respectively. Deprotonation of **12** by LDA [21], addition of DMF, and reduction with NaBH₄ gave 88% of the C(8)-CH₂OH nucleoside **16** that was desilylated to **17** (80%). Dimethoxytritylation of **16** gave **18** (89%) that was desilylated to **19** (80%). The alcohol **19** was converted to the phosphoramidite **1** in the usual way [22]; **1** is stable at –15° for several months.

The configuration of **3** was established by X-ray crystal-structure analysis of **3**·MeOH³⁾ (*Fig. 2*). The erythrofuranosyl moiety adopts the ²T₃ conformation, similarly to the conformation deduced for **3** in D₂O solution [10]. HO–C(3') forms a weak intramolecular H-bond to O–C(2') (distance O(2')...O(3') 2.77 Å); intermolecular H-bonds are found between HO–C(3') and N(1), H–N(6) and O–C(2'), MeOH and

³⁾ The crystallographic data have been deposited with the *Cambridge Crystallographic Data Centre* (deposition No. CCDC-158435 (**3**), CCDC-158436 (**4**), and CCDC-158437 (**25**)). Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge, CB21EZ, UK (fax: +44(1223)336033; e-mail: deposit@ccdc.cam.ac.uk).

N(7), and between HO–C(2') and MeOH. The nucleobase adopts the *syn*-conformation.

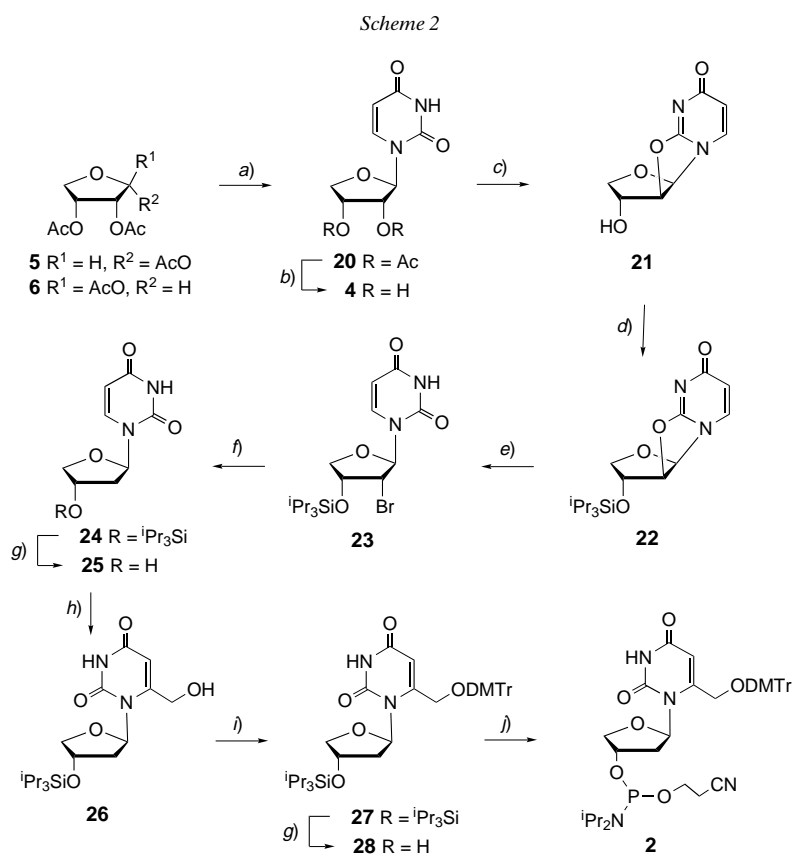
The structures of **8** and **9** were assigned on the basis of the ¹H-NMR spectra of **8/9** (2:1). The signals of H–C(2') and H–C(3') of **8/9** were assigned on the basis of their multiplicity and coupling constants (see *Table 3*). In CDCl₃, H–C(2') of the major compound appears as a *dt* with $J(2',3') = 5.0$, $J(1',2') = 4.1$, and $J(\text{H},\text{OH}) = 5.3$ Hz. Upon addition of D₂O, this signal collapsed to a *dd*, while the H–C(3') *q* remained unchanged, evidencing that the major compound possesses structure **8**. The broad H–C(3') signal of the minor compound (**9**) collapsed to a *ddd* upon addition of D₂O, while the H–C(2') *t* remained unchanged. The structure determination was corroborated by the ¹H-NMR spectra of a 2:1 mixture of the thiocarbamates **10/11**. The H–C(2') and H–C(3') signals of **10/11** were assigned on the basis of their multiplicity and coupling constants (see *Table 3*). Carbamoylation of O–C(2') of the major compound **8**, leading to **10**, induced a 1.6-ppm downfield shift for H–C(2'), while carbamoylation of O–C(3') of the minor compound **9**, leading to **11**, resulted in a 1.67-ppm downfield shift for H–C(3'). As expected, H–C(1') of **12** appears as a *t* at 6.46 ppm with $J(1',2') = 6.6$ Hz (see *Table 3* in *Exper. Part*). The H_{pro-S}–C(2') of **12** gives rise to a *ddd* at 2.97 ppm, while H_{pro-R}–C(2') appears at 2.60 ppm as a *dddd* (W coupling to H–C(4'), $J(2',4') = 1.0$ Hz). The signals of H–C(1'), H–C(2'), H–C(3'), and H–C(4') of **13** were assigned on the basis of irradiation experiments. H–C(1') of **13** resonates at 6.00 ppm as a *d*, $J(1',2') = 1.6$ Hz, H_{pro-S}–C(3') of **13** at 2.32 ppm as a *dtd*, and H_{pro-R}–C(3') at 2.12 ppm as a *dddd*. As a rule, the shift from the *anti*- to the *syn*-conformation is indicated by a downfield shift for H_{pro-S}–C(2') and an upfield shift for C(2') [23]. The significant downfield shifts for H_{pro-S}–C(2') ($\Delta\delta = 0.18$ – 0.30 ppm) and the upfield shifts for C(2') ($\Delta\delta = -1.1$ to -2.6 ppm) of the nucleosides **16**, **18**, **19**, and the phosphoramidite **1**, as compared to the corresponding signals of **12**, evidence that all C(8)-substituted nucleosides adopt predominantly the *syn*-conformation (*Table I*).

The ratio $J(1',2'_{\text{pro-S}})/J(3'_{\text{pro-R}},4')$ is considered a measure of the position of the equilibrium between the *S*- (sugar-pucker ${}^3E-{}^2T_1$) and *N*-conformers (sugar-pucker ${}^3T_2-{}^3T_4$) [24]. The $J(1',2'_{\text{pro-S}})$ (6.6 Hz) and $J(3',4')$ (≤ 2.1 Hz) values (*Table I*) evidence a predominance of the *S*-conformer of the nucleosides **1**, **12**, **14**, **16**, **18**, and **19**.

Table 1. ¹H- and ¹³C-NMR Shift Differences $\Delta\delta(\text{H}-\text{C}(2'))$, and $\Delta\delta(\text{C}(2'))$ [ppm] for the Nucleosides **16**, **18**, **19**, and **1** as Compared to **12**, and for **24**, **26**–**28**, and for **2** as Compared to **24**, and Ratio of *S/N*-Conformers of these Nucleosides, as Deduced from $J(1',2')/J(3',4')$ [Hz]

Nucleoside	$\Delta\delta(\text{H}-\text{C}(2'_{\text{pro-S}}))$	$\Delta\delta(\text{C}(2'))$	$J(1',2')$	$J(3',4')$	Ratio <i>S/N</i>
12	–	–	6.6	2.1	76/24
16	0.18	–1.1	6.6	<1.0	>87/13
18	0.23	–1.1	6.6	<1.0	>87/13
19	0.23	–1.8	6.6	<1.0	>87/13
1	0.30	–2.6	6.6	<1.0	>87/13
24	–	–	6.5	2.2	75/25
26	0.79	–2.0	6.6	<1.0	>87/13
27	0.74	–2.1	7.0	<1.0	>88/12
28	0.71	–3.1	7.0	<1.0	>88/12
2	0.77	–4.1	6.5	<1.0	>87/13

To prepare the phosphoramidite **2** (Scheme 2), the anomeric mixture **5/6** was treated with uracil in the presence of BSA and SnCl₄ in MeCN at 60° (cf. [17][18]). We obtained 80% of the β-D-configured nucleoside **20**, which was deacetylated with 2M NaOH to yield 80% of **4**. To deoxygenate **4** at C(2'), we planned to convert **4** to the corresponding 2,2'-anhydronucleoside [25], to cleave the C(2')–O bond with LiBr and BF₃·Et₂O [26], and to reductively debrominate the resulting 2'-bromo-2'-deoxynucleoside. The 2,2'-anhydro-1-(β-D-threofuranosyl)uracil (**21**) was prepared from **4** by treatment with diphenyl carbonate and NaHCO₃ in DMF [25]. However, all attempts to convert **21** to the 2'-deoxynucleoside **25** failed. Silylation of the 2,2'-anhydronucleoside **21** with a 5-fold excess of ⁱPr₃SiOTf yielded **22** (61% from **4**). Treatment of **22** with LiBr and BF₃·Et₂O gave the 2'-bromo-2'-deoxynucleoside **23** (88%) that was debrominated in 97% yield with Bu₃SnH and AIBN. The resulting 2'-deoxynucleoside



a) 1. Uracil, *N,O*-bis(trimethylsilyl)acetamide (BSA); 2. SnCl₄; 80%. *b*) 1. NaOH, H₂O/MeOH/THF 5:4:1; 2. NH₄Cl; 80%. *c*) 1. Diphenyl carbonate, DMF; 2. NaHCO₃. *d*) ⁱPr₃SiOTf, pyridine, DMF; 62% (from **4**). *e*) LiBr, BF₃·Et₂O, 1,4-dioxane; 88%. *f*) Bu₃SnH, AIBN, toluene; >97%. *g*) Bu₄NF·3 H₂O, THF; **25** (88%), **28** (96%). *h*) 1. LDA/THF; 2. DMF; 3. AcOH; 4. NaBH₄/EtOH; 51%. *i*) DMTrCl, ⁱPr₂NEt, DMAP, CH₂Cl₂; 77%. *j*) (NCCH₂CH₂O)P(ⁱPr₂N)Cl, CH₂Cl₂, ⁱPr₂NEt; 69% of 2 diastereoisomers (2:1).

24 was desilylated to **25** (80%). Hydroxymethylation of **24** to **26** was accomplished in 52% yield similarly as described above for **12**. Dimethoxytritylation of **26** to **27** and desilylation of **27** yielded 73% of the alcohol **28**, which was converted to the phosphoramidite **2** in the usual way (70%); at -15° , **2** is stable for several months.

The β -D-configuration of 1-(β -D-erythrofuranosyl)uracil (**4**) was established by X-ray crystal-structure analysis (Fig. 3, a).

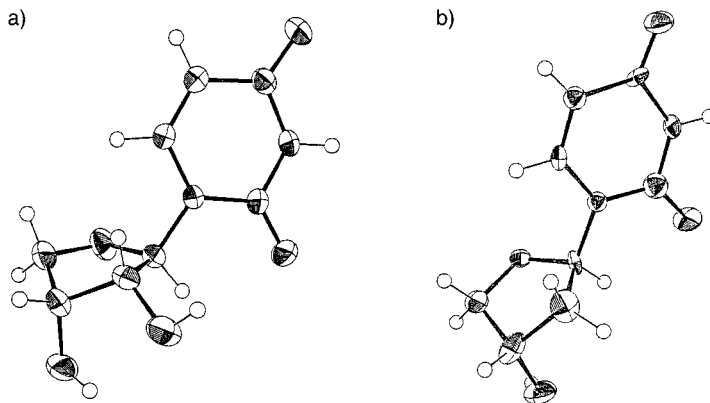


Fig. 3. Crystal structures of a) **4** and b) **25**

Similarly to N^6 -benzoyl-9-(β -D-erythrofuranosyl)adenine (**3**), the erythrofuranosyl moiety of **4** adopts the ${}_3E$ -conformation. There is a (weak?) intramolecular H-bond between HO–C(3') and O–C(2') (distance O(2') \cdots O(3') 2.69 Å), while intermolecular H-bonds are found between HO–C(2') and O–C(2'), HO–C(3') and O–C(4'), and H–N(3) and O–C(4'). Unlike **3**, the uracil derivative **4** adopts the *anti*-conformation.

The H-atoms H–C(1'), H–C(2'), H–C(3'), H–C(5), and H–C(6) of the 2,2'-anhydronucleoside **21** (Table 5 in *Exper. Part*) resonate at almost the same field as the corresponding signals of 2,2'-anhydrouridine ($\Delta\delta = \pm 0.04$ ppm) [27]. As observed for 5'-protected 2,2'-anhydrouridines [28] [29], C(2), C(4), C(5), C(1'), and C(2') of **21** are shifted significantly downfield, as compared to **4**: C(4) and C(2) give rise to two *s* at 178.7 and 164.2 ppm (**4**: two *s* at 163.5 and 151.0 ppm), C(5) resonates as a *d* at 112.1 ppm (**4**: *d* at 102 ppm), and C(1') and C(2') as two *d* at 93.4 and 90.6 ppm (**4**: two *d* at 88.8 and 74.1 ppm). Bromination of **22** to **23** is indicated by a strong upfield shift for C(2') of **23** ($\delta = 52.4$ as compared to $\delta = 74.1$ ppm in **4**), and the $[M + H]^+$ peak in the FAB-MS of **23** shows the typical isotope pattern for Br compounds.

As expected, H–C(1') of **24** appears at 6.15 ppm as a *t* with $J(1',2') = 6.5$ Hz. The H_{pro-R} -C(2') of **24** gives rise to a *ddd* at 2.50, and H_{pro-S} -C(2') to a *dt* at 2.11 ppm. The ratio $J(1',2')/J(3',4')$ indicates that the 2-deoxy-D-erythrofuranosyl moiety of **24** adopts predominantly the *S*-conformation (Table 1).

The structure of **25** was established by X-ray crystal-structure analysis (Fig. 3, b). Two symmetrically independent molecules in the elementary cell form a dimer with intermolecular H-bonds between H–N(3) and O–C(4), and between H–N(3) and O–C(2). Both independent molecules adopt the 2T_3 - and the *anti*-conformations.

Similarly to the adenine series, the downfield shift for $H_{pro-S}-C(2')$ ($\Delta\delta = 0.71 - 0.79$ ppm) and the upfield shift for $C(2')$ of **2** and **26–28**, as compared to **24** ($\Delta\delta = -2.0$ to -4.1 ppm), show the predominance of the *syn*-conformation for the $C(6)$ -substituted nucleosides (*Table 1*). The equilibrium between the *S*- and the *N*-conformers of **2** and **26–28** is clearly in favour of the *S*-conformer ($> 88/12$), as indicated by $J(1',2'_{pro-S})$ and $J(3',4')$ values of **2** and **26–28** (*Table 1*).

2. Solid-Phase Synthesis of 14-Mer DNA Strands Containing deA* or deU*, and of the Complementary DNA Strands. *Table 2* shows the 14-mer DNA strands **29** and **30** possessing a random sequence with an internal dA or dT unit flanked by dG or dC units. Replacement of the internal dA of **29** by deA* (derived from **1**) led to the modified 14-mer **31**, while replacement of the internal dT of **30** by deU* (from **2**) led to the modified 14-mer **32**. The solid-phase synthesis of the oligomers **29–32** was carried out on a DNA-synthesizer using essentially the protocol for the synthesis of pRNA [30]. The coupling time for the attachment of **1**, **2**, and the DNA phosphoramidites following the incorporation of **1** and **2** was arbitrarily set to 30 min; it led to coupling yields of 82%⁴⁾ for **1** and *ca.* 80–90% for the DNA-phosphoramidites following **1**, and of *ca.* 90% for **2** (99% for all other couplings). Longer coupling times did not improve yields. The DNA 14-mers **29** and **30** were deprotected and cleaved off the solid support by ammonolysis with saturated aqueous $NH_3/MeOH$ 1:1 at 50° within 20 h, while **31** was deprotected under the same conditions within 3.5 h. Prolonged heating of the modified DNA strand **32** (incorporating deU*) at 50° led to degradation; this strand was deprotected at 35°. The crude products were analyzed by RP-HPLC. The DNA strands **29** and **30** were pure, the modified tetradecamer **32** was *ca.* 70–80% pure, while crude **31** contained three major products A, B, and C in a *ca.* 1:2:1 ratio. The products were purified by RP-HPLC, desalted, and analyzed by RP-HPLC and MALDI-TOF mass spectroscopy; only fraction C (from crude **31**) exhibited a molecular ion in agreement with structure **31**⁵⁾. The yield of the oligomers, as determined from the detritylation assay, and their molecular masses are shown in *Table 2*.

Table 2. Total Yields, and Calculated and Experimental Molecular Masses of the Oligodeoxynucleotides **29–32**

Sequence	Total yield [%] ^{a)}	Calc. mass	Exper. mass (MALDI-TOF)
29 5'-d(CGTAAGCTCGATAG)-3'	94	4287.8	4287.4
30 5'-d(CTATCGAGCTTACG)-3'	99	4238.8	4239.1
31 5'-d(CTATCGeA*GCTTACG)-3'	27	4238.8	4237.7
32 5'-d(CGTAAGCeU*CGATAG)-3'	75	4273.8	4276.6

^{a)} Calculated from the yield of the coupling steps, as determined by the detritylation assay.

3. Pairing Studies. The modified 14-mer **31** (*Table 2*) was hybridized with the complementary 14-mer **30** at a concentration of (2+2) μM and at pH 7 (10 mM NaH_2PO_4/Na_2HPO_4 in H_2O) in 0.1M aqueous NaCl. The modified duplex **31·30**

⁴⁾ Each chain elongation cycle began with an acid promoted cleavage of the dimethoxytrityl group. The efficiency of each coupling step was calculated by UV-determination of the amount of $DMTr^+$ ($\epsilon = 70000$ l·mol⁻¹·cm⁻¹).

⁵⁾ In agreement with this assignment, the fractions A and B did not pair with **30**.

showed a decrease of the melting temperature by 8°, as compared to the non-modified duplex **29**·**30** (see Fig. 4 for the melting curves).

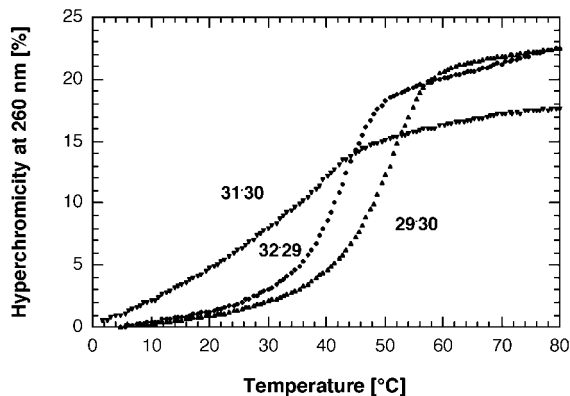


Fig. 4. Temperature-dependent UV spectra ('melting curves') of the duplexes **29**·**30**, **31**·**30**, and **32**·**29**

Similarly, the modified 14-mer **32** (Table 2) incorporating one deU* unit was hybridized with the complementary 14-mer **29** at a concentration of (2 + 2) μM and at pH 7 (10 mM $\text{NaH}_2\text{PO}_4/\text{Na}_2\text{HPO}_4$ in H_2O) in 0.1M aqueous NaCl. Again, the modified duplex **32**·**29** melted 7° lower than the non-modified DNA duplex **29**·**30** (see Fig. 4 for the melting curves). Since the modified 14-mer **32** is degraded during prolonged heating, the melting process for the duplex **32**·**29** was irreversible.

Similarly as reported for the incorporation of the 2-deoxyribose-derived units dA* and dU* [5], the introduction of the 2-deoxy-D-erythrose-derived units deA* and deU* into a DNA duplex causes a decrease in melting temperature by 7–8°, as compared to the unmodified duplex. A comparison of the duplex destabilization imposed by dA*, dU*, deA*, and deU* shows that this destabilization is independent of the CH_2OH substituent at C(4').

We thank Dr. B. Bernet for checking the experimental part, and the Swiss National Science Foundation and F. Hoffmann-La Roche AG, Basel for generous support.

Experimental Part

General. Solvents were distilled before use: THF from K/benzophenone, toluene from Na, CH_2Cl_2 , DMF, pyridine, 1,4-dioxane, and Et_3NPr_2 from CaH_2 . Reactions were run under Ar. Qual. TLC: precoated silica-gel plates (Merck silica gel 60 F_{254}); detection by spraying with 'mostain' (400 ml of 10% aq. H_2SO_4 , 20 g of $(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} \cdot \text{H}_2\text{O}$, 0.4 g of $\text{Ce}(\text{SO}_4)_2$) and heating. Flash chromatography (FC): silica gel Merck 60 (0.04–0.063 mm). Reversed-Phase High-Performance Liquid Chromatography (RP-HPLC): analytical RP column: Macherey-Nagel; LiChroCart®250-4 HPLC Cartridge filled with LiChrospher®100, RP-18e (5 μm); prep. RP column: Macherey-Nagel, Hibar® RT250-10, customized packing: LiChrospher®100, RP-18e (5 μm), eluents: 0.1M aq. Et_3NHOAc , MeCN, H_2O . Optical rotations: 1-dm cell at 25° and 589 nm. FT-IR: 1–2% soln. in the indicated solvent. ^1H -, ^{13}C -, and ^{31}P -NMR: at 200, 300, or 400 MHz, 50 or 75 MHz, or 121 MHz, resp. MS: fast-atom bombardment (FAB), matrix-assisted laser-desorption ionization (MALDI), high-resolution (HR). NOBA: 3-nitrobenzyl alcohol.

N^6 -Benzoyl-9-(2',3'-di-O-acetyl- β -D-erythrofuranosyl)adenine (**7**). A suspension of **5/6** (1:4) [11–13] (4.16 g, 16.9 mmol) and N^6 -benzoyladenine (4.4 g, 18.4 mmol) in MeCN (18 ml) was treated dropwise with N,O -bis(trimethylsilyl)acetamide (10 ml, 41 mmol) at 60°. The mixture was stirred for 30 min, treated with SnCl_4

(8 ml, 68 mmol), stirred for 20 min, poured into a stirred mixture of AcOEt (300 ml) and sat. aq. NaHCO₃ (300 ml), and stirred for 15 min. The org. layer was dried (Na₂SO₄) and evaporated. FC (AcOEt/MeOH 100 : 1) gave **7** (5.7 g, 80%). Colourless powder. *R*_f(AcOEt/hexane 6 : 1) 0.40. [α]_D²⁵ = –90.1 (*c* = 0.75, CHCl₃). IR (CHCl₃): 3375*w* (br.), 3007*m*, 1750*s*, 1709*s*, 1612*s*, 1587*s*, 1504*m*, 1480*s*, 1469*s*, 1374*s*, 1072*s*. ¹H-NMR (200 MHz, CDCl₃): see Table 3; additionally, 9.22 (br. *s*, NH); 8.05–7.95 (*m*, 2 arom. H); 7.63–7.40 (*m*, 3 arom. H); 2.16 (*s*, AcO); 2.05 (*s*, AcO). ¹³C-NMR (50 MHz, CDCl₃): see Table 4; additionally, 170.0, 169.5 (2*s*, 2 O–C=O); 164.9 (*s*, N–C=O); 133.0 (*s*); 132.9 (*d*); 128.9 (*d*, 2 C); 128.0 (*d*, 2 C); 20.7 (*q*, Me); 20.4 (*q*, Me). FAB-MS (NOBA): 426 (69, *M*⁺), 427 (25, [*M* + H]⁺). HR-MALDI-MS: 448.123 (C₂₀H₁₉N₅NaO₆, [*M* + Na]⁺; calc. 448.123).

Table 3. Selected ¹H-NMR Chemical Shifts [ppm] and Coupling Constants [Hz] for the Nucleosides **7**–**13**, **16**–**19**, and **1** in CDCl₃, for **3** in (D₆)DMSO, and for **14** and **15** in CD₃OD

	7	3	8	9	10	11	12	13
H–C(1')	6.1–6.2	5.97	5.94	5.90	6.33	6.00	6.46	6.00
H–C(2' _{pro-S})	6.1–6.2	4.86	4.83	5.56	6.42	5.90	2.97	5.10
H–C(2' _{pro-R})	–	–	–	–	–	–	2.60	–
H _a –C(3')	5.74	4.28	5.00	4.45	5.47	6.12	4.93	2.32
H _b –C(3')	–	–	–	–	–	–	–	2.12
H _a –C(4')	4.73	4.40	4.50	4.58	4.62	4.88	4.43	4.42
H _b –C(4')	4.17	3.84	4.00	4.15	4.00	4.37	4.03	4.32
H–C(2)	8.75	8.75	8.76	8.80	8.71	8.76	8.82	8.80
H–C(8)	8.10	8.68	8.08	8.00	8.35	8.45	8.13	8.05
<i>J</i> (1',2')	^{a)}	6.6	4.2	5.3	4.1	6.2	6.6	1.6
<i>J</i> (2',2')	–	–	–	–	–	–	13.7	–
<i>J</i> (2' _{pro-S} ,3')	5.1	4.1	5.0	5.0	4.4	4.4	6.6	4.4
<i>J</i> (2' _{pro-R} ,3')	–	–	–	–	–	–	2.5	2.4
<i>J</i> (3',3')	–	–	–	–	–	–	–	13.0
<i>J</i> (3',4'a)	4.1	3.7	5.0	3.4	5.0	3.0	4.0	8.4, 4.0
<i>J</i> (3',4'b)	2.1	1.2	4.1	<1.0	5.0	<1.0	2.1	8.7, 6.2
<i>J</i> (4'a,4'b)	10.4	9.1	9.3	10.0	9.1	11.5	9.1	8.4
	14	15	16	17	18	19	1	
H–C(1')	6.57	6.10	6.50	6.68	6.47	6.37	6.46	
H–C(2' _{pro-S})	2.95	4.84	3.15	3.25	3.20	3.20	3.27	
H–C(2' _{pro-R})	2.58	–	2.39	2.50	2.43	2.35	2.60, 2.51	
H _a –C(3')	4.77	2.40	4.95	4.82	4.94	4.87	4.85	
H _b –C(3')	–	2.10	–	–	–	–	–	
H _a –C(4')	4.42	4.42	4.36	4.57	4.55	4.50	4.53	
H _b –C(4')	3.96	4.28	3.89	3.96	3.92	3.92	4.08, 4.04	
H _a –C(10)	–	–	4.90	4.95	4.55	4.52	4.53, 4.50	
H _b –C(10)	–	–	4.90	4.90	4.40	4.45	4.45, 4.42	
H–C(2)	8.70	8.70	8.68	8.66	8.80	8.76	8.78	
H–C(8)	8.50	8.42	–	–	–	–	–	
<i>J</i> (1',2')	6.6	1.9	6.6	6.8	6.6	6.6	6.6	
<i>J</i> (2',2')	14.0	–	13.7	14.0	13.3	14.0	13.3	
<i>J</i> (2' _{pro-S} ,3')	5.6	5.6	6.8	5.6	6.6	6.0	6.4	
<i>J</i> (2' _{pro-R} ,3')	1.2	3.1	1.0	1.3	<1.0	<1.0	<1.0	
<i>J</i> (3'a,3'b)	–	14.0	–	–	–	–	–	
<i>J</i> (3',4'a)	3.7	8.4, 4.3	4.0	3.7	3.3	3.7	3.3	
<i>J</i> (3',4'b)	1.2	8.4, 7.1	<1.0	<1.0	<1.0	<1.0	<1.0	
<i>J</i> (4'a,4'b)	9.3	8.4	8.7	9.0	8.7	9.3	9.3	
<i>J</i> (10 _a ,10 _b)	–	–	^{a)}	14.3	11.8	11.8	12.1	

^{a)} Not assigned.

Table 4. Selected ^{13}C -NMR Chemical Shifts [ppm] for **7**–**13**, **16**–**19**, and **1** in CDCl_3 , for **14** and **15** in CD_3OD , and for **3** in $(D_6)\text{DMSO}$

	C(1')	C(2')	C(3')	C(4')	C(10)	C(2)	C(4)	C(5)	C(6)	C(8)
7	87.3	74.2	71.6	72.7	–	151.8	149.9	124.1	152.9	142.1
3	88.0	74.5	70.4	74.0	–	151.9	150.6	126.3	152.6	144.0
8	91.4	75.2	71.7	74.7	–	152.9	150.2	124.1	151.5	142.5
9	91.2	75.9	71.9	74.7	–	153.2	150.2	124.1	151.5	143.1
10	87.7	83.4	70.4	74.4	–	152.6	150.3	124.2	151.7	142.3
11	91.0	74.3	82.4	72.0	–	152.7	150.3	124.2	151.7	143.0
12	86.2	41.7	72.6	77.3	–	152.8	149.9	124.2	151.8	142.2
13	92.9	76.4	33.8	69.3	–	152.5	149.6	123.8	151.2	141.3
14	87.2	41.7	73.0	78.0	–	153.0	151.3	125.6	153.1	144.8
15	93.8	76.7	33.8	70.5	–	153.0	151.0	125.6	152.8	144.0
16	85.6	40.6	73.0	77.6	57.8	154.8	149.3	122.0	153.0 ^{a)}	152.4 ^{a)}
17	87.2	40.7	73.3	78.2	58.3	156.5	150.5	124.2	154.5 ^{a)}	152.9 ^{a)}
18	86.1	40.6	73.2	77.8	59.7	152.0	149.5	122.5	152.7 ^{a)}	152.6 ^{a)}
19	85.7	39.9	72.7	77.4	59.5	152.0	149.4	122.5	152.6 ^{a)}	152.5 ^{a)}
1	85.9	39.2	74.4	76.5	59.9	152.2	149.7	122.7	152.7 ^{a)}	152.6 ^{a)}
	85.8	39.0	74.2	76.2		152.1	149.6			

^{a)} Assignment may be interchanged.

N^6 -Benzoyl-9-(β -D-erythrofuranosyl)adenine·MeOH (**3**). At 23°, a soln. of **7** (5.2 g, 12.2 mmol) in THF (190 ml), MeOH (152 ml), and H₂O (38 ml) was treated with 2M NaOH in H₂O (25 ml, 50 mmol), stirred at 23° for 10 min, treated with NH₄Cl (5.26 g, 98.3 mmol), and evaporated after addition of SiO₂ (46 g). The resulting powder was dried at 0.1 mbar for 14 h, filled in a column, and eluted with CHCl₃/MeOH 10:1. The colourless crystals obtained from a few pooled fractions were submitted to X-ray analysis. Evaporation of the residual fractions gave **3** (3.2 g, 78%). Colourless powder. R_f (CHCl₃/MeOH 10:1) 0.20. IR (KBr): 3407s (br.), 3237s (br.), 1685s, 1608m, 1572s, 1518s, 1466m, 1428m, 1396m, 1325m, 1294m, 1248m, 1194m, 1146m, 1107m, 1070s, 1025s. ¹H-NMR (200 MHz, (D₆)DMSO): see Table 3; additionally, 11.17 (br. s, NH); 8.06–7.97 (m, 2 arom. H); 7.69–7.47 (m, 3 arom. H); 5.56 (d, $J=6.2$, HO–C(2')); 5.29 (d, $J=3.7$, HO–C(3')); 4.01 (q, $J=5.5$, MeOH); 3.16 (d, $J=5.5$, MeOH). ¹³C-NMR (50 MHz, (D₆)DMSO): see Table 4; additionally, 165.9 (s, C=O); 133.6 (s); 132.6 (d); 128.7 (d, 4 C); 48.6 (q, MeOH). FAB-MS (NOBA): 342 (100, [M+H]⁺).

N^6 -Benzoyl-9-(3'-O-(triisopropylsilyl)- β -D-erythrofuranosyl)adenine (**8**) and N^6 -Benzoyl-9-(2'-O-(triisopropylsilyl)- β -D-erythrofuranosyl)adenine (**9**). A soln. of **3** (3.2 g, 9.4 mmol) in pyridine (4.1 ml, distilled from CaH₂) and DMF (64 ml, distilled from CaH₂) was treated with ³Pr₃SiOTf (3.2 ml, 12 mmol) at 23°, stirred for 4 h, treated with ³Pr₃SiOTf (3 ml, 11 mmol), stirred for 4 h at 23°, and evaporated. FC (AcOEt/hexane 6:1) gave **8/9** 1:1 (4.3 g, 92%). A soln. of this mixture in 150 ml of Et₃N was treated with SiO₂ (60 g), stirred at 23° for 16 h, and evaporated. FC (CH₂Cl₂/MeOH 10:1) gave **8/9** 2:1. Colourless foam.

Data of **8/9**: R_f (AcOEt/hexane 6:1) 0.41. $[\alpha]_D^{25} = -58.4$ ($c=1.135$, CHCl₃, **8/9** 2:1). IR (CHCl₃, **8/9** 2:1): 3411w (br.), 3003m, 2947s, 2893m, 2869s, 1708s, 1671m, 1612s, 1585s, 1503m, 1457s, 1387m, 1354m, 1328m, 1141m, 1112m. ¹H-NMR (300 MHz, CDCl₃, **8/9** 2:1): see Table 3; additionally, 9.13 (br. s, 0.33 H), 9.10 (br. s, 0.67 H) (NH); 8.05–7.95 (m, 2 arom. H); 7.65–7.40 (m, 3 arom. H); 3.45 (d, $J=5.6$, 0.67 H, exch., HO–C(2') of **8**); 3.02–2.96 (m, 0.33 H, exch., HO–C(3') von **9**); 1.20–0.80 (m, (Me₂CH)₃Si). ¹³C-NMR (75 MHz, CDCl₃, **8/9** 2:1): see Table 4; additionally, 165.9 (s, C=O); 134.0 (s); 133.1 (d); 129.0 (d, 2 C); 128.4 (d, 2 C); 17.9–17.5 (4q, (Me₂CH)₃Si); 12.4–12.0 (3d, (Me₂CH)₃Si). FAB-MS (NOBA, **8/9** 2:1): 498 (100, [M+H]⁺). HR-MALDI-MS (**8/9** 2:1): 498.254 (C₂₅H₃₆N₅O₄Si, [M+H]⁺; calc. 498.254); 520.236 (C₂₅H₃₅N₅NaO₄Si, [M+Na]⁺; calc. 520.236).

N^6 -Benzoyl-9-[2'-O-(imidazol-1-yl)thiocarbonyl]-3'-O-(triisopropylsilyl)- β -D-erythrofuranosyl)adenine (**10**) and N^6 -Benzoyl-9-(3'-O-(imidazol-1-yl)thiocarbonyl)-2'-O-(triisopropylsilyl)- β -D-erythrofuranosyl)adenine (**11**). A soln. of **8/9** 2:1 (2 g, 4 mmol) in CH₂Cl₂ (20 ml) was treated with 1,1'-(thiocarbonyl)diimidazole (4.6 g, 26 mmol), stirred at 23° for 3 d, extracted once with H₂O and brine, and dried (Na₂SO₄). Evaporation gave a yellow oil (3.8 g), consisting of **10/11** (2:1) and 1,1'-(thiocarbonyl)diimidazole. It was used without

further purification. R_f (AcOEt/hexane 6:1, **10/11** 2:1) 0.33. $^1\text{H-NMR}$ (300 MHz, CDCl_3 , **10/11** 2:1): see *Table 3*; additionally, 9.40 (br. s, 0.67 H), 9.37 (br. s, 0.33 H) (NH); 8.20–7.80 (*m*, 3 arom. H); 7.80–7.30 (*m*, 3 arom. H); 7.20–6.95 (*m*, 2 arom. H); 1.20–0.80 (*m*, $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 4*; additionally, 183.0, 182.9 (2s, C=O); 165.3, 165.1 (2s, C=O); 137.0 (*d*); 133.7 (*s*); 133.6 (*s*); 132.8–128.0 (several *d*); 118.1 (*d*); 17.8–17.4 (several *q*, $(\text{Me}_2\text{CH})_3\text{Si}$), 12.1, 11.9 (2*d*, $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA, **10/11** 2:1): 608 (63, $[M+H]^+$).

N^6 -Benzoyl-9-[2'-deoxy-3'-O-(triisopropylsilyl)- β -D-glycero-tetrofuranosyl]adenine (**12**) and N^6 -Benzoyl-9-[3'-deoxy-2'-O-(triisopropylsilyl)- β -D-glycero-tetrofuranosyl]adenine (**13**). A suspension of **10/11** 2:1 (3.8 g) in toluene (20 ml) and a mixture of Bu_3SnH (6 ml, 22.6 mmol) and AIBN (3.5 g, 21 mmol) in toluene (10 ml) were each degassed. The suspension of **10/11** was heated to 110°, treated with the soln. of AIBN and Bu_3SnH in 6 portions over 40 min, and stirred at 110° for 2 h. The soln. was diluted with CHCl_3 (200 ml), extracted once with H_2O and brine, dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane 1:1) gave **12** (850 mg, 44% from **8/9**) and **13** (385 mg, 20% from **8/9**) as colourless foams.

Data of 12: R_f (AcOEt/hexane 6:1) 0.51. $[\alpha]_D^{25} = -40.8$ ($c = 1.07$, CHCl_3). IR (CHCl_3): 3406w (br.), 3003m, 2946m, 2892w, 2868m, 1708m, 1612s, 1583s, 1503m, 1457s, 1386w, 1346w, 1328m, 1071s. $^1\text{H-NMR}$ (200 MHz, CDCl_3): see *Table 3*; additionally, 9.00 (br. s, NH); 8.10–8.00 (*m*, 2 arom. H); 7.65–7.50 (*m*, 3 arom. H); 1.20–1.00 (*m*, $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 4*; additionally, 165.2 (*s*, C=O); 134.0 (*s*); 133.1 (*d*); 129.2 (*d*, 2 C); 128.3 (*d*, 2 C); 17.9 (*q*, $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (*d*, $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 482 (77, $[M+H]^+$). HR-FAB-MS: 482.2585 ($\text{C}_{25}\text{H}_{36}\text{N}_5\text{O}_3\text{Si}$, $[M+H]^+$; calc. 482.2587).

Data of 13: R_f (AcOEt/hexane 6:1) 0.57. $[\alpha]_D^{25} = -51.1$ ($c = 1.065$, CHCl_3). IR (CHCl_3): 3403w (br.), 3002m, 2946m, 2892w, 2868m, 1707m, 1612s, 1584m, 1501m, 1455s, 1327w. $^1\text{H-NMR}$ (300 MHz, CDCl_3): see *Table 3*; additionally, 9.00 (br. s, NH); 8.07–7.98 (*m*, 2 arom. H); 7.65–7.45 (*m*, 3 arom. H); 1.20–1.00 (*m*, $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 4*; additionally, 165.0 (*s*, C=O); 133.7 (*s*); 132.7 (*d*); 128.7 (*d*, 2 C); 128.0 (*d*, 2 C); 17.9 (*q*, $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (*d*, $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 482 (55, $[M+H]^+$). HR-MALDI-MS: 504.240 ($\text{C}_{25}\text{H}_{35}\text{N}_5\text{NaO}_3\text{Si}$, $[M+Na]^+$; calc. 504.240).

N^6 -Benzoyl-9-(2'-deoxy- β -D-glycero-tetrofuranosyl)adenine (**14**). A soln. of **12** (20 mg, 0.042 mmol) in THF (2 ml) was treated with 1M Bu_4NF in THF (0.24 ml, 0.24 mmol) at 23° for 3 h and evaporated after addition of SiO_2 (0.5 g). The resulting powder was dried at 0.1 mbar for 14 h and filled in a column. Elution ($\text{CHCl}_3/\text{MeOH}$ 20:1) gave **14** (10 mg, 74%) as colourless powder. R_f ($\text{CHCl}_3/\text{MeOH}$ 10:1) 0.29. $^1\text{H-NMR}$ (300 MHz, CD_3OD): see *Table 3*; additionally, 8.10–8.00 (*m*, 2 arom. H); 7.70–7.50 (*m*, 3 arom. H). $^{13}\text{C-NMR}$ (75 MHz, CD_3OD): see *Table 4*; additionally, 168.0 (*s*, C=O); 135.0 (*s*); 134.0 (*d*); 130.0 (*d*, 2 C); 129.8 (*d*, 2 C). FAB-MS (NOBA): 326 (56, $[M+H]^+$), 348 (16, $[M+Na]^+$).

N^6 -Benzoyl-9-(3'-deoxy- β -D-glycero-tetrofuranosyl)adenine (**15**). A soln. of **13** (100 mg, 0.208 mmol) in THF (2 ml) was treated with 1M Bu_4NF in THF (0.62 ml, 0.62 mmol) at 23° for 3 h and evaporated after addition of SiO_2 (0.5 g). The resulting powder was dried at 0.1 mbar for 14 h and filled in a column. Elution ($\text{CHCl}_3/\text{MeOH}$ 20:1) gave **15** (60 mg, 89%) as colourless powder. R_f ($\text{CHCl}_3/\text{MeOH}$ 10:1) 0.42. $^1\text{H-NMR}$ (300 MHz, CD_3OD): see *Table 3*; additionally, 8.10–8.00 (*m*, 2 arom. H); 7.70–7.50 (*m*, 3 arom. H). $^{13}\text{C-NMR}$ (75 MHz, CD_3OD): see *Table 4*; additionally, 168.0 (*s*, C=O); 135.0 (*s*); 134.0 (*d*); 129.9 (*d*, 2 C); 129.5 (*d*, 2 C). FAB-MS (NOBA): 326 (65, $[M+H]^+$).

N^6 -Benzoyl-9-(2-deoxy-3-O-(triisopropylsilyl)- β -D-erythrofuranosyl)-8-(hydroxymethyl)adenine (**16**). At –70°, a soln. of **12** (800 mg, 1.663 mmol) in THF (20 ml) was treated dropwise with 2M LDA in THF (4.2 ml, 8.4 mmol), stirred at –70° for 2 h, treated with DMF (3.3 ml, 42.8 mmol), stirred at –70° for 2.5 h, treated with AcOH (1.4 ml), allowed to warm to 23°, and diluted with EtOH (30 ml). The resulting soln. was treated with NaBH_4 (220 mg, 5.8 mmol) for 25 min and evaporated. The residue was taken up in CH_2Cl_2 (100 ml), washed once with H_2O (50 ml) and brine (50 ml), dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane 6:1) gave **16** (748 mg, 88%). Colourless foam. R_f (AcOEt/hexane 6:1) 0.34. $[\alpha]_D^{25} = -40.4$ ($c = 1.095$, CHCl_3). IR (CHCl_3): 3404m (br.), 3006m, 2945s, 2892m, 2868s, 1706s, 1655m, 1614s, 1585s, 1462s, 1436s, 1384m, 1336s, 1072s. $^1\text{H-NMR}$ (300 MHz, CDCl_3): see *Table 3*; additionally, 9.60 (br. s, NH); 8.10–7.90 (*m*, 2 arom. H); 7.60–7.40 (*m*, 3 arom. H); 4.15 (br. s, exch., OH); 1.20–0.80 (*m*, $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 4*; additionally, 165.5 (*s*, C=O); 134.2 (*s*); 132.8 (*d*); 129.0 (*d*, 2 C); 128.3 (*d*, 2 C); 17.9 (*q*, $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (*d*, $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 512 (35, $[M+H]^+$). HR-MALDI-MS: 512.268 ($\text{C}_{26}\text{H}_{38}\text{N}_5\text{O}_4\text{Si}$, $[M+H]^+$; calc. 512.269), 534.251 ($\text{C}_{26}\text{H}_{37}\text{N}_5\text{NaO}_4\text{Si}$, $[M+Na]^+$; calc. 534.251).

N^6 -Benzoyl-9-(2-deoxy- β -D-erythrofuranosyl)-8-(hydroxymethyl)adenine (**17**). A soln. of **16** (20 mg, 0.039 mmol) in THF (1 ml) was treated with 1M Bu_4NF in THF (0.12 ml, 0.12 mmol) at 23° for 3 h and evaporated after addition of SiO_2 (0.4 g). The resulting powder was dried at 0.1 mbar for 14 h and filled in a column. Elution ($\text{CHCl}_3/\text{MeOH}$ 20:1 to 10:1) gave **17** (11 mg, 80%) as colourless powder. R_f ($\text{CHCl}_3/\text{MeOH}$

6:1) 0.51. ¹H-NMR (300 MHz, CD₃OD): see Table 3; additionally, 8.10–8.00 (*m*, 2 arom. H); 7.70–7.50 (*m*, 3 arom. H). ¹³C-NMR (75 MHz, CD₃OD): see Table 4; additionally, 168.3 (*s*, C=O); 135.2 (*s*); 134.0 (*d*); 129.8 (*d*, 2 C); 129.5 (*d*, 2 C). FAB-MS (NOBA): 356 (78, [*M* + H]⁺), 378 (41, [*M* + Na]⁺).

*N*⁶-Benzoyl-9-[2'-deoxy-3'-O-(triisopropylsilyl)-β-D-glycero-tetrofuranosyl]-8-[(4,4'-dimethoxytrityloxy)methyl]adenine (**18**). A soln. of **16** (748 mg, 1.46 mmol), EtN(i-Pr)₂ (1.03 ml, 6 mmol) and DMAP (50 mg, 0.4 mmol) in CH₂Cl₂ (15 ml) was treated with DMTrCl (2 g, 5.91 mmol) at 0°. The mixture was stirred at 23° for 16 h, and evaporated. The residue was diluted with CH₂Cl₂, washed once with H₂O and brine, dried (Na₂SO₄), and evaporated. FC (AcOEt/hexane 1:2) gave **18** (1056 mg, 89%). Colourless foam. *R*_f(AcOEt/hexane 2:1) 0.61. [*α*]_D²⁵ = –15.2 (*c* = 2.32, CHCl₃). IR (CHCl₃): 3409w (br.), 3007m, 2960m, 2867w, 1706m, 1610s, 1584m, 1509s, 1463m, 1447m, 1428m, 1070m, 1034m. ¹H-NMR (200 MHz, CDCl₃): see Table 3; additionally, 9.06 (br. *s*, NH); 8.10–8.00 (*m*, 2 arom. H); 7.69–7.20 (*m*, 12 arom. H); 6.92–6.80 (*m*, 4 arom. H); 3.78 (*s*, 2 MeO); 1.20–1.00 (*m*, (Me₂CH)₃Si). ¹³C-NMR (75 MHz, CDCl₃): see Table 4; additionally, 165.0 (*s*, C=O); 159.2 (*s*, 2 C); 144.4 (*s*); 135.6 (*s*); 135.3 (*s*); 134.4 (*s*); 132.8 (*d*); 130.2–127.2 (several *d*); 113.3 (*d*, 4 C); 87.9 (*s*, Ar₃C); 55.3 (*q*, 2 MeO); 18.0 (*q*, (Me₂CH)₃Si), 12.0 (*d*, (Me₂CH)₃Si). FAB-MS (NOBA): 814 (38, [*M* + H]⁺), 303 (100, DMTr⁺). HR-MALDI-MS: 836.382 (C₄₇H₅₅N₅NaO₆Si, [*M* + Na]⁺; calc. 836.381).

*N*⁶-Benzoyl-9-[2'-deoxy-β-D-glycero-tetrofuranosyl]-8-[(4,4'-dimethoxytrityloxy)methyl]adenine (**19**). A soln. of **18** (976 mg, 1.2 mmol) in THF (15 ml) was treated dropwise with 1M Bu₄NF in THF (3.55 ml, 3.55 mmol) at 23° for 1.5 h and evaporated. FC (AcOEt/hexane 2:1) gave **19** (600 mg, 76%). Colourless foam. *R*_f(AcOEt/hexane 2:1) 0.13. [*α*]_D²⁵ = –12.5 (*c* = 1.69, CHCl₃). IR (CHCl₃): 3405m (br.), 3007s, 2839w, 1707m, 1613s, 1585m, 1509s, 1462s, 1448s, 1428s, 1409m, 1336s, 1302s, 1065s, 1036s. ¹H-NMR (300 MHz, CDCl₃): see Table 3; additionally, 9.20 (br. *s*, NH); 8.05–7.95 (*m*, 2 arom. H); 7.60–7.10 (*m*, 12 arom. H); 6.85–6.75 (*m*, 4 arom. H); 3.75 (*s*, 2 MeO); 2.60–2.40 (br. *s*, exch., HO–C(3')). ¹³C-NMR (50 MHz, CDCl₃): see Table 4; additionally, 164.9 (*s*, C=O); 159.0 (*s*, 2 C); 144.3 (*s*); 135.3 (*s*); 135.2 (*s*); 134.2 (*s*); 132.9 (*d*); 130.2–127.3 (several *d*); 113.6 (*d*, 4 C); 88.0 (*s*, Ar₃C); 55.3 (*q*, 2 MeO). FAB-MS (NOBA): 659 (2, [*M* + H]⁺), 303 (100, DMTr⁺). HR-FAB-MS (NOBA): 658.2663 (C₃₈H₃₆N₅O₆, [*M* + H]⁺; calc. 658.2666).

*N*⁶-Benzoyl-9-(2'-deoxy-β-D-glycero-tetrofuranosyl)-8-[(4,4'-dimethoxytrityloxy)methyl]adenine 3'-[(2-Cyanoethyl)diisopropylphosphoramidite] (**1**). A soln. of **19** (395 mg, 0.601 mmol) and EtN(i-Pr)₂ (2 ml) in CH₂Cl₂ (20 ml) was treated with 2-cyanoethyl diisopropylchlorophosphoramidite (184 mg, 0.77 mmol) at 23°, stirred for 3 h, and evaporated. FC (AcOEt/hexane 2:1) gave **1** (413 mg, 80%, 2 diastereoisomers (1:1)). Colourless foam. *R*_f(AcOEt/hexane 2:1, 2 diastereoisomers (1:1)) 0.34; 0.45. IR (CHCl₃, 2 diastereoisomers (1:1)): 3404w (br.), 3007m, 2970m, 2935m, 1707m, 1613s, 1585m, 1509s, 1462m, 1448m, 1428w, 1408m, 1365m, 1337m, 1063m, 1036m. ¹H-NMR (300 MHz, CDCl₃, 2 diastereoisomers (1:1)): see Table 3; additionally, 9.00 (br. *s*, NH); 8.10–7.95 (*m*, 2 arom. H); 7.70–7.20 (*m*, 12 arom. H); 6.90–6.80 (*m*, 4 arom. H); 3.90–3.50 (*m*, OCH₂CH₂N, (Me₂CH)₂N); 3.78 (*s*, 2 MeO); 2.63 (*t*, *J* = 6.3), 2.55 (*t*, *J* = 6.1) (CH₂CN); 1.30–1.10 (*m*, (Me₂CH)₂N). ¹³C-NMR (75 MHz, CDCl₃, 2 diastereoisomers (1:1)): see Table 4; additionally, 165.1 (*s*, N–C=O); 159.2 (*s*, 2 C); 144.5, 144.4 (2*s*, 1 C); 135.6, 135.5, 135.4, 134.4 (several *s*, 3 C); 132.9–127.5 (several *d*); 117.8 (*s*, CN); 113.7 (*d*, 4 C); 88.1 (*s*, Ar₃C); 58.5 (*dt*, ²*J*(C,P) = 19.0, OCH₂CH₂CN); 55.4 (*q*, 2 MeO); 43.4 (*dd*, ²*J*(C,P) = 12.0, (Me₂CH)₂N); 24.8–24.5 (several *q*, (Me₂CH)₂N); 20.5–20.3 (*m*, OCH₂CH₂CN). ³¹P-NMR (121.5 MHz, CDCl₃): 148.9; 148.6. FAB-MS (NOBA): 858 (49, [*M* + H]⁺), 303 (100, DMTr⁺). HR-MALDI-MS: 880.356 (C₄₇H₅₂N₇NaO₇P, [*M* + Na]⁺; calc. 880.356).

1-(2',3'-Di-O-acetyl-β-D-erythrofuranosyl)uracil (**20**). A suspension of **5/6** 1:4 (1.8 g, 7.3 mmol) and uracil (0.92 g, 8.2 mmol) in MeCN (18 ml) was treated dropwise with *N*,*O*-bis(trimethylsilyl)acetamide (4.32 ml, 17.7 mmol) at 60°, stirred at 60° for 30 min, and treated with SnCl₄ (3.47 ml, 29.5 mmol). The mixture was stirred at 60° for 20 min, poured into a stirred mixture of AcOEt (300 ml) and sat. aq. NaHCO₃ (300 ml), and stirred for 15 min. The org. layer was dried (Na₂SO₄) and evaporated. FC (AcOEt/hexane 10:1) gave **20** (1.74 g, 80%). Colourless powder. *R*_f(AcOEt/hexane 6:1) 0.38. ¹H-NMR (200 MHz, CDCl₃): see Table 5; additionally, 9.48 (br. *s*, NH); 2.14 (*s*, AcO); 2.09 (*s*, AcO). ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 170.4, 170.2 (2*s*, 2 O–C=O); 20.7, 20.5 (2*q*, 2 Me).

1-(β-D-Erythrofuranosyl)uracil (**4**). At 23°, a soln. of **20** (1.1 g, 3.7 mmol) in a mixture of THF (94 ml), MeOH (75 ml), and H₂O (19 ml) was treated with 2M NaOH in H₂O (12.4 ml, 25 mmol), stirred at 23° for 8 min, treated with NH₄Cl (2.6 g, 48.6 mmol), treated with SiO₂ (22.8 g), and evaporated. The resulting colourless powder was dried at 0.1 mbar for 14 h, filled in a column, and eluted (CHCl₃/MeOH 6:1). The colourless crystals obtained from a few pooled fractions were submitted to X-ray analysis. Evaporation of the residual fractions gave **4** (0.63 g, 80%). Colourless crystals. *R*_f(CHCl₃/MeOH 6:1) 0.16. [*α*]_D²⁵ = –41.2 (*c* = 0.9, H₂O). IR (KBr): 3385s (br.), 3186s, 1693s, 1470m, 1404s, 1262m, 1130m, 1065m. ¹H-NMR (300 MHz, (D₆)DMSO): see

Table 5. Selected ¹H-NMR Chemical Shifts [ppm] and Coupling Constants [Hz] for **20**, **22–24**, **26–28**, and **2** in CDCl₃, for **4** in (D₆)DMSO, and for **21** and **25** in D₂O

	2	4	20	22	23	24	26	27	28	21	25
H–C(1')	5.87	5.70	5.86	6.27	5.90	6.15	6.11	5.90	5.85	6.56	6.18
H–C(2' _{pro-S})	2.88	4.18	5.5–5.6	5.13	4.64	2.11	2.90	2.85	2.82	5.45	2.19
H–C(2' _{pro-R})	2.32, 2.25	–	–	–	–	2.50	2.23	2.18	2.12	–	2.40
H–C(3')	4.72	4.08	5.4–5.5	4.64	4.53	4.65	4.83	4.73	4.65	4.70	4.53
H _a –C(4')	4.34	4.21	4.52	4.04	4.35	4.20	4.42	4.38	4.34	4.16	4.25
H _b –C(4')	3.4–3.9	3.67	4.06	3.80	4.02	3.87	3.83	3.73	3.73	3.93	3.88
H _a –C(7)	3.98, 3.97	–	–	–	–	–	4.52	3.96	4.02	–	–
H _b –C(7)	3.98, 3.97	–	–	–	–	–	4.45	3.96	3.95	–	–
H–C(5)	5.76, 5.73	5.60	5.77	6.06	5.76	5.73	5.80	5.80	5.80	6.22	5.69
H–C(6)	–	7.65	7.22	7.38	7.20	7.32	–	–	–	7.94	7.61
J(1',2')	6.5	6.2	5.3	5.0	5.6	6.5	6.6	7.0	7.0	5.3	7.0
J(2',2')	13.7	–	–	–	–	14.0	12.8	13.0	13.4	–	14.0
J(2' _{pro-S} ,3')	6.5	4.4	^{a)}	<1.0	4.7	6.2	6.5	6.5	6.5	<1.0	5.3
J(2' _{pro-R} ,3')	^{a)}	–	–	–	–	2.3	<1.0	<1.0	<1.0	–	1.6
J(3',4'a)	4.4	4.0	4.5	<1.0	4.4	4.2	4.4	4.1	4.0	<1.0	3.7
J(3',4'b)	<1.0	2.2	2.0	2.5	3.7	2.2	<1.0	<1.0	<1.0	2.8	1.5
J(4'a,4'b)	8.7	9.1	10.3	10.6	9.0	9.0	8.5	8.4	9.0	11.2	9.3
J(5,6)	–	8.1	8.1	7.5	8.1	8.1	–	–	–	7.5	8.1
J(7 _a ,7 _b)	^{a)}	–	–	–	–	–	14.7	^{a)}	13.4	–	–

^{a)} Not assigned.Table 6. Selected ¹³C-NMR Chemical Shifts [ppm] for **20**, **22–24**, **26–28**, and **2** in CDCl₃, for **4** in (D₆)DMSO, and for **21** and **25** in D₂O

	C(1')	C(2')	C(3')	C(4')	C(7)	C(6)	C(2)	C(4)	C(5)
20	89.9	73.8	71.1	72.4	–	140.7	150.7	163.4	103.4
4	88.8	74.1	70.0	73.6	–	141.8	151.0	163.5	102.0
21	93.4 ^{a)}	90.6 ^{a)}	75.7 ^{b)}	75.7 ^{b)}	–	141.7	164.2	178.7	112.1
22	90.3 ^{a)}	87.7 ^{a)}	75.0 ^{b)}	74.0 ^{b)}	–	135.2	160.3	172.0	110.8
23	95.4	52.4	71.7	74.8	–	141.4	150.1	163.8	102.9
24	88.0	42.4	72.0	77.2	–	140.0	150.8	164.0	102.6
25	88.4	41.9	72.3	77.5	–	142.3	152.1	166.0	102.6
26	87.7	40.4	73.6	78.7	61.2	156.0	151.2	164.0	101.6
27	88.1	40.3	73.5	78.5	62.8	153.5	151.2	163.8	102.5
28	88.1	39.3	73.0	78.0	62.5	153.3	151.2	163.6	102.6
2	88.2	38.4	74.5	79.8	62.8	153.3	150.6	162.6	102.6
		38.3	74.3	79.6			150.5		102.5

^{a)} ^{b)} Assignments may be interchanged.

Table 5; additionally, 11.22 (br. s, NH); 5.34 (*d*, *J* = 6.2, *exch.*, HO–C(2')); 5.07 (*d*, *J* = 3.7, *exch.*, HO–C(3')).
¹³C-NMR (50 MHz, (D₆)DMSO): see Table 6. FAB-MS (NOBA): 215 (100, [*M* + H]⁺).

2,2'-Anhydro-1-(β-D-threofuranosyl)uracil (**21**). A suspension of **4** (1 g, 4.67 mmol) and diphenyl carbonate (1.29 g, 0.61 mmol) in DMF (25 ml) was stirred at 80° for 30 min, treated with NaHCO₃ (390 mg) at 80°, stirred at 150° for 2 h, and evaporated. The residue was diluted with AcOEt and H₂O. Evaporation of the aq. layer gave 998 mg of a colourless powder containing **21** and Na salts. R_f(AcOEt/MeOH 3:2) 0.57. [*α*]_D²⁵ = –44 (*c* = 1.05, H₂O). IR (KBr): 2923*m*, 1662*s*, 1620*m*, 1533*s*, 1482*s*, 1272*w*, 1241*w*, 1190*w*, 1113*w*, 1062*m*, 1041*m*. ¹H-NMR (300 MHz, D₂O): see Table 5. ¹³C-NMR (75 MHz, D₂O): see Table 6. FAB-MS (NOBA): 219 (100, [*M* + Na]⁺), 197 (24, [*M* + H]⁺). HR-MALDI-MS: 219.037 (C₈H₈N₂NaO₄, [*M* + Na]⁺; calc. 219.038).

2,2'-Anhydro-1-[3'-O-(triisopropylsilyl)- β -D-threofuranosyl]uracil (22). A soln. of **21** (1 g, 5.1 mmol) in DMF (16 ml) and pyridine (12.5 ml) was treated with $^i\text{Pr}_3\text{SiOTf}$ (7 ml, 26 mmol), stirred at 23° for 24 h, and evaporated. FC (AcOEt/MeOH 30:1) gave **22** (790 mg, 61.5% from **4**). Colourless foam. $R_f(\text{AcOEt/MeOH } 10:1)$ 0.37. $[\alpha]_D^{25} = -40.6$ ($c = 0.9$, CHCl_3). IR (CHCl_3): 2944s, 1672s, 1644s, 1544m, 1468s, 1387w, 1134m, 1072m, 1042m. $^1\text{H-NMR}$ (200 MHz, CDCl_3): see *Table 5*; additionally, 1.20–0.90 (m , $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 6*; additionally, 18.0 (q , $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (d , $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 353 (100, $[M + H]^+$), 705 (8, $[2M + H]^+$). HR-MALDI-MS: 353.189 ($\text{C}_{17}\text{H}_{29}\text{N}_2\text{O}_4\text{Si}$, $[M + H]^+$; 353.189).

1-[2'-Bromo-2'-deoxy-3'-O-(triisopropylsilyl)- β -D-erythrofuranosyl]uracil (23). A suspension of **22** (670 mg, 1.9 mmol) and LiBr (215 mg, 2.47 mmol) in 1,4-dioxane (20 ml) was heated to 60° and treated dropwise with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (0.62 ml, 4.9 mmol). The mixture was stirred at 60° for 6 h, and taken to dryness. The residue was diluted with CHCl_3 (50 ml), washed once with H_2O (20 ml) and brine (20 ml), dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane 1:1) gave **23** (720 mg, 88%). Colourless foam. $R_f(\text{AcOEt/hexane } 1:1)$ 0.34. $[\alpha]_D^{25} = -29.9$ ($c = 1.15$, CHCl_3). IR (CHCl_3): 3391w (br.), 2940m, 2868m, 1695s, 1633w, 1457m, 1384m, 1143m, 1108m, 1068m, 1017m. $^1\text{H-NMR}$ (300 MHz, CDCl_3): see *Table 5*; additionally, 8.60 (br. s, NH); 1.15–1.03 (m , $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 6*; additionally, 18.0 (q , $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (d , $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 433, 435 (23, 24; $[M + H]^+$). HR-MALDI-MS: 375.173 ($\text{C}_{17}\text{H}_{28}\text{N}_2\text{NaO}_4\text{Si}$, $[M - \text{HBr} + \text{Na}]^+$; calc. 375.172).

1-[2'-Deoxy-3'-O-(triisopropylsilyl)- β -D-glycero-tetrofuranosyl]uracil (24). A suspension of **23** (696 mg, 1.61 mmol) in toluene (10 ml) and a mixture of Bu_3SnH (0.43 ml, 1.62 mmol) and AIBN (100 mg, 0.61 mmol) in toluene (5 ml) were each degassed. The suspension of **23** was heated to 80° and treated with the soln. of AIBN and Bu_3SnH over 20 min. The mixture was heated to 110°, stirred for 30 min, diluted with CHCl_3 (200 ml), washed with H_2O and brine, dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane 1:1) gave **24** (556 mg, 97%). Colourless foam. $R_f(\text{AcOEt/hexane } 1:1)$ 0.30. IR (CHCl_3): 3391w (br.), 2946m, 1690s, 1633w, 1461m, 1390m, 1135m, 1073m. $^1\text{H-NMR}$ (300 MHz, CDCl_3): see *Table 5*; additionally, 9.67 (br. s, NH); 1.17–0.94 (m , $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 6*; additionally, 18.0 (q , $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (d , $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 355 (18, $[M + H]^+$). HR-FAB-MS: 355.2052 ($\text{C}_{17}\text{H}_{31}\text{N}_2\text{O}_4\text{Si}$, $[M + H]^+$; calc. 355.2053).

1-[2'-Deoxy- β -D-glycero-tetrofuranosyl]uracil (25). A soln. of **24** (91 mg, 0.258 mmol) in THF (2 ml) was treated with $1\text{M Bu}_4\text{NF} \cdot 3 \text{H}_2\text{O}$ in THF (0.76 ml, 0.76 mmol) at 23° for 3 h, and evaporated after addition of SiO_2 (0.5 g). The resulting powder was dried at 0.1 mbar for 14 h, filled in a column, and eluted ($\text{CHCl}_3/\text{MeOH } 20:1$). Evaporation gave **25** (45 mg, 88%) as white powder. Recrystallization in MeOH gave colourless crystals used for X-ray analysis. $R_f(\text{CHCl}_3/\text{MeOH } 10:1)$ 0.28. $^1\text{H-NMR}$ (300 MHz, CD_3OD): see *Table 5*. $^{13}\text{C-NMR}$ (75 MHz, CD_3OD): see *Table 6*. FAB-MS (NOBA): 199 (28, $[M + H]^+$).

1-[2'-Deoxy-3'-O-(triisopropylsilyl)- β -D-glycero-tetrofuranosyl]-6-(hydroxymethyl)uracil (26). At -70° , 2M LDA in THF/heptane/ethylbenzene (*FLUKA*; 1.63 ml, 3.26 mmol) was treated with a soln. of **24** (230 mg, 0.652 mmol) in THF (4 ml). The mixture was stirred at -70° for 30 min, treated with DMF (1.7 ml, 16.5 mmol), stirred at -70° for 2.5 h, and treated with AcOH (0.55 ml). The mixture was allowed to warm to 23°, diluted with EtOH (4 ml), and treated with NaBH_4 (83 mg, 2.17 mmol) for 25 min. After evaporation, a soln. of the residue in CH_2Cl_2 was washed with H_2O and brine, dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane 4:1) gave **26** (128 mg, 51%). Colourless foam. $R_f(\text{AcOEt/hexane } 1:1)$ 0.15. $[\alpha]_D^{25} = -82.2$ ($c = 0.4$, CHCl_3). IR (CHCl_3): 3390m (br.), 2946s, 2867s, 1697s, 1602m, 1462m, 1377m, 1094s, 1015s. $^1\text{H-NMR}$ (300 MHz, $\text{CDCl}_3/\text{CD}_3\text{OD}$): see *Table 5*; additionally, 1.16–0.94 (m , $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 6*; additionally, 18.0 (q , $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (d , $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 385 (21, $[M + H]^+$).

1-[2'-Deoxy-3'-O-(triisopropylsilyl)- β -D-glycero-tetrofuranosyl]-6-[4,4'-dimethoxytrityloxy)methyl]uracil (27). A soln. of **26** (117 mg, 0.305 mmol), $\text{EtN}(\text{i-Pr})_2$ (0.24 ml, 1.39 mmol) and DMAP (10 mg, 0.4 mmol) in CH_2Cl_2 (5 ml) was treated with DMTrCl (413 mg, 1.26 mmol) at 0°, stirred at 23° for 16 h, and evaporated. A soln. of the residue in CH_2Cl_2 (50 ml) was washed once with H_2O (20 ml) and brine (20 ml), dried (Na_2SO_4), and evaporated. FC (AcOEt/hexane/ $\text{Et}_3\text{N } 1:1:0.02$) gave **27** (161 mg, 77%). Colourless foam. $R_f(\text{AcOEt/hexane } 1:1)$ 0.36. $[\alpha]_D^{25} = -25$ ($c = 0.1$, CHCl_3). IR (CHCl_3): 3389m (br.), 2975m, 1695s, 1607m, 1509s, 1457w, 1045m. $^1\text{H-NMR}$ (300 MHz, CDCl_3): see *Table 5*; additionally, 9.75 (br. s, NH); 7.50–7.10 (m , 9 arom. H); 6.93–6.70 (m , 4 arom. H); 3.80 (s , 2 MeO); 1.15–0.75 (m , $(\text{Me}_2\text{CH})_3\text{Si}$). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3): see *Table 6*; additionally, 159.4 (s , 2 C); 144.3 (s); 135.5 (s); 135.1 (s); 130.1–127.6 (several d); 113.6, 113.5 (2d, 4 C); 88.6 (s , Ar_3C); 55.4 (q , 2 MeO); 18.0 (q , $(\text{Me}_2\text{CH})_3\text{Si}$); 12.0 (d , $(\text{Me}_2\text{CH})_3\text{Si}$). FAB-MS (NOBA): 686 (7, $[M + H]^+$); 303 (100, DMTr^+).

1-[2'-Deoxy- β -D-glycero-tetrofuranosyl]-6-[4,4'-dimethoxytrityloxy)methyl]uracil (28). A soln. of **27** (161 mg, 0.235 mmol) in THF (6 ml) was treated dropwise at 23° with $1\text{M Bu}_4\text{NF}$ in THF (0.7 ml, 0.7 mmol)

for 1.5 h, and evaporated. FC (AcOEt/hexane/Et₃N 10:1:0.02) gave **28** (120 mg, 96%). Colourless foam. R_f (AcOEt/hexane 10:1) 0.24. $[\alpha]_D^{25} = -17.9$ ($c = 0.9$, CHCl₃). IR (CHCl₃): 3390m (br.), 2961m, 1692s, 1608m, 1584w, 1509s, 1464s, 1410m, 1378m, 1302m, 1066s, 1035s. ¹H-NMR (300 MHz, CDCl₃): see Table 5; additionally, 9.60 (br. s, NH); 7.50–7.20 (m, 9 arom. H); 6.90–6.70 (m, 4 arom. H); 3.75 (s, 2 MeO). ¹³C-NMR (75 MHz, CDCl₃): see Table 6; additionally, 159.2 (s, 2 C); 144.2 (s); 135.2 (s, 2 C); 130.1–127.7 (several d); 114.0 (d, 4 C); 88.2 (s, Ar₃C); 55.3 (q, 2 MeO). FAB-MS (NOBA): 530 (12, [M+H]⁺), 303 (100, DMTr⁺). HR-FAB-MS: 530.2053 (C₃₀H₃₀N₂O₇, M⁺; calc. 530.2053).

1-(2'-Deoxy-β-D-glycero-tetrofuranosyl)-6-[4,4'-dimethoxytrityloxy)methyl]uracil 3'-[(2-Cyanoethyl)diisopropylphosphoramidite] (2). A soln. of **28** (108 mg, 0.204 mmol) and EtN(i-Pr)₂ (0.12 ml, 0.69 mmol, distilled from CaH₂) in CH₂Cl₂ (2 ml) was treated with 2-cyanoethyl diisopropylchlorophosphoramidite (46 μl, 0.206 mmol) at 23°, stirred for 3 h, and evaporated. FC (AcOEt/hexane 2:1) gave **2** (102 mg, 69%, 2 diastereoisomers (2:1)). Colourless foam. R_f (AcOEt/hexane 1:1, 2 diastereoisomers (2:1)) 0.26. IR (CHCl₃, 2 diastereoisomers (2:1)): 3391w (br.), 2967s, 2929s, 1693s, 1608s, 1509s, 1463s, 1378s, 1064m, 1035m. ¹H-NMR (300 MHz, CDCl₃, 2 diastereoisomers (2:1)): see Table 5; additionally, 8.30–8.20 (br. s, NH); 7.45–7.20 (m, 9 arom. H); 6.90–6.80 (m, 4 arom. H); 3.90–3.45 (m, Me₂CH)₂N, OCH₂CH₂CN, H–C(4''); 3.80 (s, 2 MeO); 2.62 (t, $J = 6.5$, 0.67 H), 2.52 (t, $J = 6.5$, 1.33 H) (CH₂CN); 1.36–1.00 (m, (Me₂CH)₂N). ¹³C-NMR (75 MHz, CDCl₃, 2 diastereoisomers (2:1)): see Table 6; additionally, 159.3 (s, 2 C); 144.2, 144.1 (2s, 1 C); 135.2, 135.0 (2s, 2 C); 130.2–127.7 (several d); 113.9 (d, 4 C); 111.3 (s, CN); 88.4 (s, Ar₃C); 58.5 (t, ²J(C,P) = 18, OCH₂CH₂CN); 55.5 (q, 2 MeO); 43.3 (dd, ²J(C,P) = 13.0, (Me₂CH)₂N); 24.8–24.4 (several q, (Me₂CH)₂N); 20.5, 20.4 (2t, OCH₂CH₂CN). ³¹P-NMR (121.5 MHz, CDCl₃, 2 diastereoisomers (2:1)): 148.7, 148.2. FAB-MS (NOBA, 2 diastereoisomers (2:1)): 513 (40, [M – OPR₂]⁺), 303 (100, DMTr⁺). HR-MALDI-MS: 535.184 (C₃₀H₂₈N₂NaO₆, [M – HOP(OCH₂CH₂CN)NⁱPr₂ + Na]⁺; calc. 535.185), 449.156 (C₁₈H₂₇N₄NaO₆P, [M – DMTrH + Na]⁺; calc. 449.157), 303.138 (C₂₁H₁₉O₂, DMTr⁺; calc. 303.139), 241.108 (C₉H₁₉N₂NaO₂P, [HOP(OCH₂CH₂CN)NⁱPr₂ + Na]⁺; calc. 241.108).

Oligonucleotide Synthesis. Oligonucleotide syntheses were performed on a *Pharmacia Gene Assembler* on a 1.3-μmol scale. The commercial phosphoramidites and the CPG solid-supports were from *Glen Research*. Solvents and reagents were prepared according to the protocol for the synthesis of pRNA [30]. Detritylation was accomplished within 2 min with 3% Cl₂CHCOOH in (CH₂Cl)₂. Couplings (0.16 ml of 0.1M phosphoramidite soln. + 0.36 ml of 0.25M 1-(benzylthio)-1H-tetrazole soln. in MeCN) were performed within 6–10 min (DNA-phosphoramidites coupled on unmodified nucleotides), or 30 min (modified phosphoramidites, and DNA-phosphoramidites coupled on modified nucleotides, resp.). Capping and oxidation were accomplished under standard conditions [31].

Deprotection and Purification of Oligonucleotides. Removal of the protecting groups and detachment from the solid support was effected in conc. aq. NH₃ soln./MeOH 1:1 (2.5 ml) at 50° within 20 h (oligomers **29** and **30**), or 3.5 h (oligomer **31**), or at 35° within 3.5 h (oligomer **32** containing dU*, see *Results and Discussion*). After filtration and evaporation of the resulting mixture, the crude oligomers were dissolved in aq. buffer soln. (0.1M AcOH, 0.1M Et₃N), purified by RP-HPLC, and desalted by RP-HPLC. Their compositions were confirmed by MALDI-TOF mass spectroscopy.

REFERENCES

- [1] S. Eppacher, N. Solladié, B. Bernet, A. Vasella, *Helv. Chim. Acta* **2000**, *83*, 1311.
- [2] H. Gunji, A. Vasella, *Helv. Chim. Acta* **2000**, *83*, 1331.
- [3] H. Gunji, A. Vasella, *Helv. Chim. Acta* **2000**, *83*, 2975.
- [4] H. Gunji, A. Vasella, *Helv. Chim. Acta* **2000**, *83*, 3229.
- [5] W. Czechtizky, A. Vasella, *Helv. Chim. Acta* **2001**, *84*, 594.
- [6] F. Mohamadi, N. G. J. Richards, W. C. Guida, R. Liskamp, M. Lipton, C. Caufield, G. Chang, T. Hendrickson, W. C. Still, *J. Comput. Chem.* **1990**, *11*, 440.
- [7] W. Saenger, 'Principles of Nucleic Acid Structure', Springer Verlag, 1984, pp. 69–78.
- [8] C. Chu, L. Teng, T. Yang, L. Kotra, F. Naguib, Q. Teng, J. Sommadossi, M. Kouni, *Tetrahedron Lett.* **1995**, *36*, 983.
- [9] J. Vastra, A. Vasella, unpublished results.
- [10] A. Serianni, P. Kline, *J. Org. Chem.* **1992**, *57*, 1772.
- [11] A. Perlin, C. Brice, *Can. J. Chem.* **1955**, *33*, 1216.
- [12] T. Storz, B. Bernet, A. Vasella, *Helv. Chim. Acta* **1999**, *82*, 2380.

- [13] E. Westerlund, R. Andersson, O. Theander, *Carbohydr. Res.* **1978**, *61*, 501.
- [14] A. Serianni, J. Snyder, *Carbohydr. Res.* **1991**, *210*, 21.
- [15] D. Critcher, S. Conolly, M. Mahon, M. Wills, *J. Chem. Soc., Chem. Commun.* **1995**, 139.
- [16] C. Demuyneck, C. Andre, J. Bolte, *Tetrahedron: Asymmetry* **1998**, *9*, 1359.
- [17] B. Vorbrüggen, B. Bennua, *Chem. Ber.* **1981**, *114*, 1279.
- [18] H. Vorbrüggen, K. Krolkiewicz, B. Bennua, *Chem. Ber.* **1981**, *114*, 1234.
- [19] J. Wagner, P. H. Verheyden, J. G. Moffatt, *J. Org. Chem.* **1974**, *39*, 24.
- [20] K. Ogilvie, G. Hakimelahi, Z. Proba, *Can. J. Chem.* **1982**, *60*, 1106.
- [21] H. Hayakawa, K. Haraguchi, H. Tanaka, T. Miyasaka, *Chem. Pharm. Bull.* **1987**, *35*, 72.
- [22] S. Pitsch, *Helv. Chim. Acta* **1997**, *80*, 2286.
- [23] L. Dudycz, R. Stolarski, R. Pless, D. Shugar, *Z. Naturforsch.* **1978**, *34c*, 359.
- [24] D. B. Davies, 'Conformations of Nucleosides and Nucleotides', in 'Progress in NMR Spectroscopy', 1978, *12*, p. 135.
- [25] A. Miah, C. B. Reese, Q. Song, Z. Sturdy, S. Neidle, I. J. Simpson, M. Read, E. Rayner, *J. Chem. Soc., Perkin Trans. 1* **1998**, 3277.
- [26] E. Kawashima, Y. Ishido, Y. Aoyama, T. Sekine, Y. Iwamoto, *Nucleosides Nucleotides* **1996**, *15*, 733.
- [27] B. P. Cross, T. Schleich, *Biopolymers* **1973**, *12*, 2381.
- [28] F. Moris, V. Gotor, *Tetrahedron* **1993**, *49*, 10089.
- [29] L. F. Garcia-Alles, J. Magdalena, V. Gotor, *J. Org. Chem.* **1996**, *61*, 6980.
- [30] S. Pitsch, S. Wendeborn, B. Jaun, A. Eschenmoser, *Helv. Chim. Acta* **1993**, *76*, 2161.
- [31] Pharmacia, 'User Manual for Gene Assembler Plus'.

Received February 23, 2001